

Fluorine battery solar container





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Why does chlorine have a higher electron affinity than fluorine?

Fluorine, though higher than chlorine in the periodic table, has a very small atomic size. This makes the fluoride anion so formed unstable (highly reactive) due to a very high charge/mass ...

Why is the carbon-fluorine bond stronger than the other haloalkanes

The partial charges on the fluorine and carbon are attractive, contributing to the unusual bond strength of the carbon-fluorine bond. The bond is labeled as " the strongest in organic ...

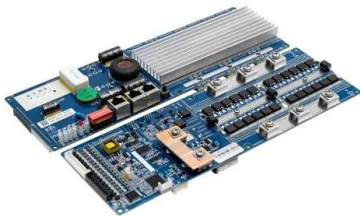


inorganic chemistry

This is shielding. Lastly, fluorine is much smaller molecule than chlorine, and the shorter distance, or radius, between the nucleus and the electron again makes it more likely to attract the ...

Is F₂ two fluoride ions or just one?

However, I'm confused with the fluoride ion having a charge of -1 and fluorine only occurring as a diatomic molecule (F_2). Is F_2 two fluoride ions or just one?



Why are there more fluoride compounds formed with xenon?

Noble gases form compounds with fluorine and oxygen only because fluorine and oxygen are the most electronegative elements. Heavier noble gases (xenon and krypton) are able to ...

Spontaneity and nature of attack of fluorine gas on aluminum

7 What is the nature of the reaction of attack of fluorine gas on aluminium metal? Is it spontaneous in nature? I have studied reactions of halogens on aluminium, but it had no information ...



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